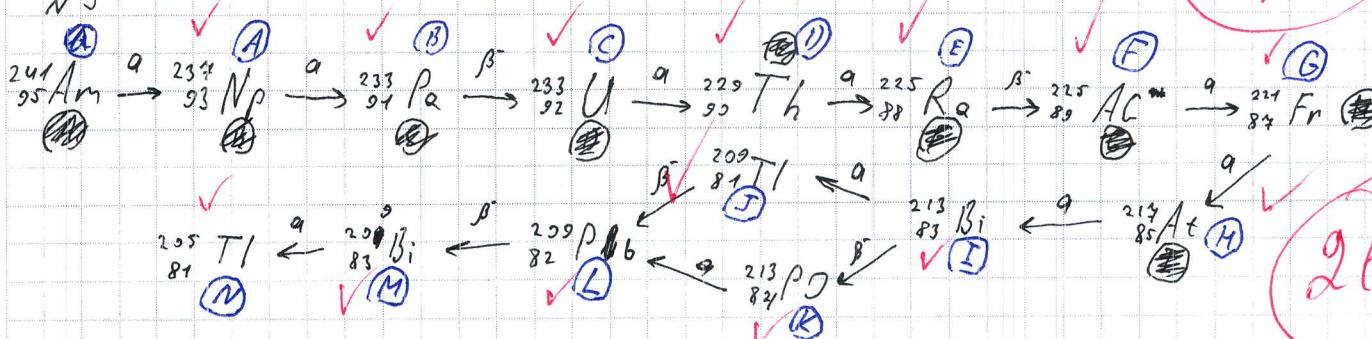


4

- | | |
|---------------|----------------------|
| 1) Золото + | 8) Рений + |
| 2) Говор + | 9) Америций + |
| 3) Осмий + | 10) Европий + |
| 4) Лютеций + | 11) Ксенон + |
| 5) Иттрий + | 12) Рентгений + |
| 6) Тантал + | 13) Модиевий Хесий + |
| 7) Огамесом + | Закордрованный |

Закорыченный элемент - ТЕХНЕЧИЙ +

5



100

2

- $$\text{All } \text{ are } W(k) = 37,84\%$$



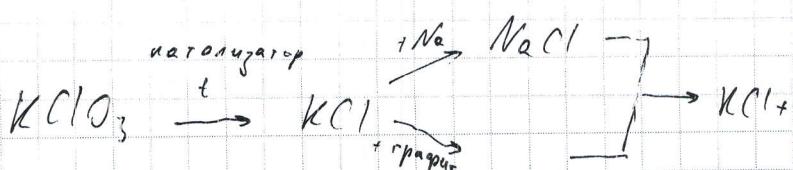
BR

20

$$W = 100 - (31,84 + 28,98) = 39,18$$

1

$$X = \text{KClO}_3 \text{ at } 100^\circ\text{C}$$



$\begin{array}{r} 3 \\ \times 2 \\ \hline 6 \end{array}$	-	$\begin{array}{r} 1 \\ \times 2 \\ \hline 2 \end{array}$	-	$\begin{array}{r} 0 \\ \times 2 \\ \hline 0 \end{array}$
$\begin{array}{r} 3 \\ \times 3 \\ \hline 9 \end{array}$	-	$\begin{array}{r} 3 \\ \times 3 \\ \hline 9 \end{array}$	-	$\begin{array}{r} 0 \\ \times 3 \\ \hline 0 \end{array}$
$\begin{array}{r} 3 \\ \times 4 \\ \hline 12 \end{array}$	-	$\begin{array}{r} 4 \\ \times 4 \\ \hline 16 \end{array}$	-	$\begin{array}{r} 14 \\ \times 4 \\ \hline 56 \end{array}$
$\begin{array}{r} 3 \\ \times 5 \\ \hline 15 \end{array}$	-	$\begin{array}{r} 5 \\ \times 5 \\ \hline 25 \end{array}$	-	$\begin{array}{r} 25 \\ \times 5 \\ \hline 125 \end{array}$

365

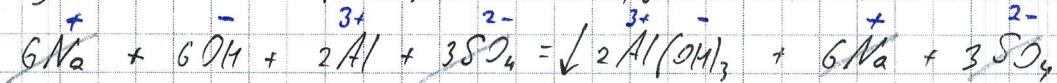
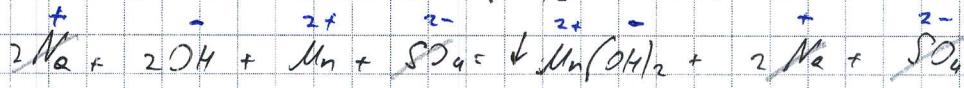
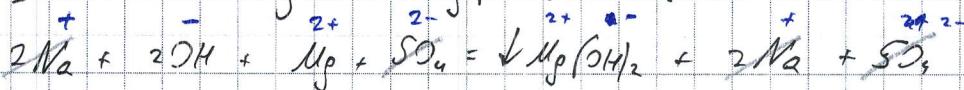
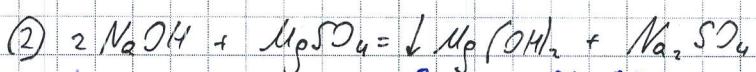
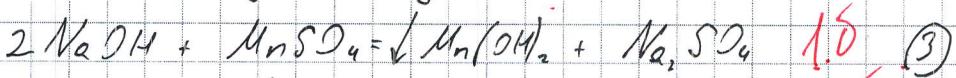
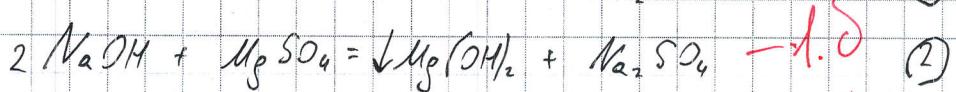
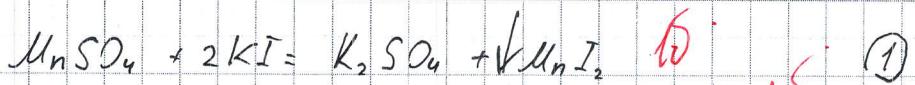
✓1

	KI	NaOH	MgSO ₄	MnSO ₄	Al ₂ (SO ₄) ₃
KI	-	-		↓	-
NaOH	-		↓	↓	↓
MgSO ₄	-	↓		-	-
MnSO ₄	↓	↓	-		-
Al ₂ (SO ₄) ₃	-	↓	-	-	

38+1 = 55
+ 45

95.

95. 95



Mg MnI₂ - светло-розовый цвет

Mg(OH)₂ - белый цвет

Mn(OH)₂ - светло-розовый цвет

Al(OH)₃ - белый цвет